

Chapter 7 – Radiation Therapy and Radiation Safety in Medicine

Questions

- Q7.2 To judge the severity you would need to know the exposure time and the RBE of the radiation used.
- Q7.3 The thin metal filters could be designed to stop low energy particles while allowing high-energy particles to be passed and measured. The thickness of the filters is determined by the energy of the particle that you need to filter. You could use filters of different thicknesses to determine the energy of the incident particles and their numbers
- Q7.7 CT scanners are used for anatomy and certainly PET and SPECT scans cannot. MRI certainly can give anatomy, but it cannot measure exposure. CT scanners can be used in-situ during a treatment, whereas PET, SPECT and MRI cannot.

Problem

- P7.1 We want some fraction of the initial source activity  $\Phi A_0$  to be deposited in the body. This activity is related to the decay constant of the reaction and to the number of nuclei, or  $\Phi A_0 = \lambda N \rightarrow N = \frac{\Phi A_0}{\lambda}$ . Each radioactive decay produces  $80.3keV$ , as a worst-case scenario, so the total energy that's deposited will be  $E_{total} = NE_{x-ray}$ . The dose to the patient

$$D = \frac{E_{total}}{m} = \frac{NE_{x-ray}}{m} = \frac{\Phi A_0}{\lambda m} E_{x-ray} = \frac{0.8 \times 1 \times 10^6 Bq}{\frac{\ln 2}{72hr \times \frac{3600s}{1hr}} \times 60kg} \times 80.3 \times 10^3 eV \times \frac{1.6 \times 10^{-19} J}{1eV}$$

$$D = 6.4 \times 10^{-5} Gy = 0.06mGy$$

- P7.2 The source activity (per unit body mass of  $60kg$ ) of potassium-40 is given as  $\frac{4630Bq}{70kg} = 66\frac{Bq}{kg}$ . Here, we don't have to worry about the half-life of the isotope since the body's supply is reasonably constantly and is continually being refreshed from the environment. For an annual dose, the exposure time,  $t = 1yr = 31.5 \times 10^6s$  and the released in every radioactive decay is given by the energy of the beta particles.  $E = 0.39MeV = 0.39 \times 10^6 eV \times \frac{1.6 \times 10^{-19} J}{1eV} = 6.24 \times 10^{-14} J$ . The dose is then:

$$D = \frac{E_{total}}{m} = 66\frac{Bq}{kg} \times 6.24 \times 10^{-14} J \times 31.5 \times 10^6 s = 1.3 \times 10^{-4} Gy$$

$$D = 0.13mGy$$

This is about one-third of the total annual dose from all internally deposited radioisotopes.