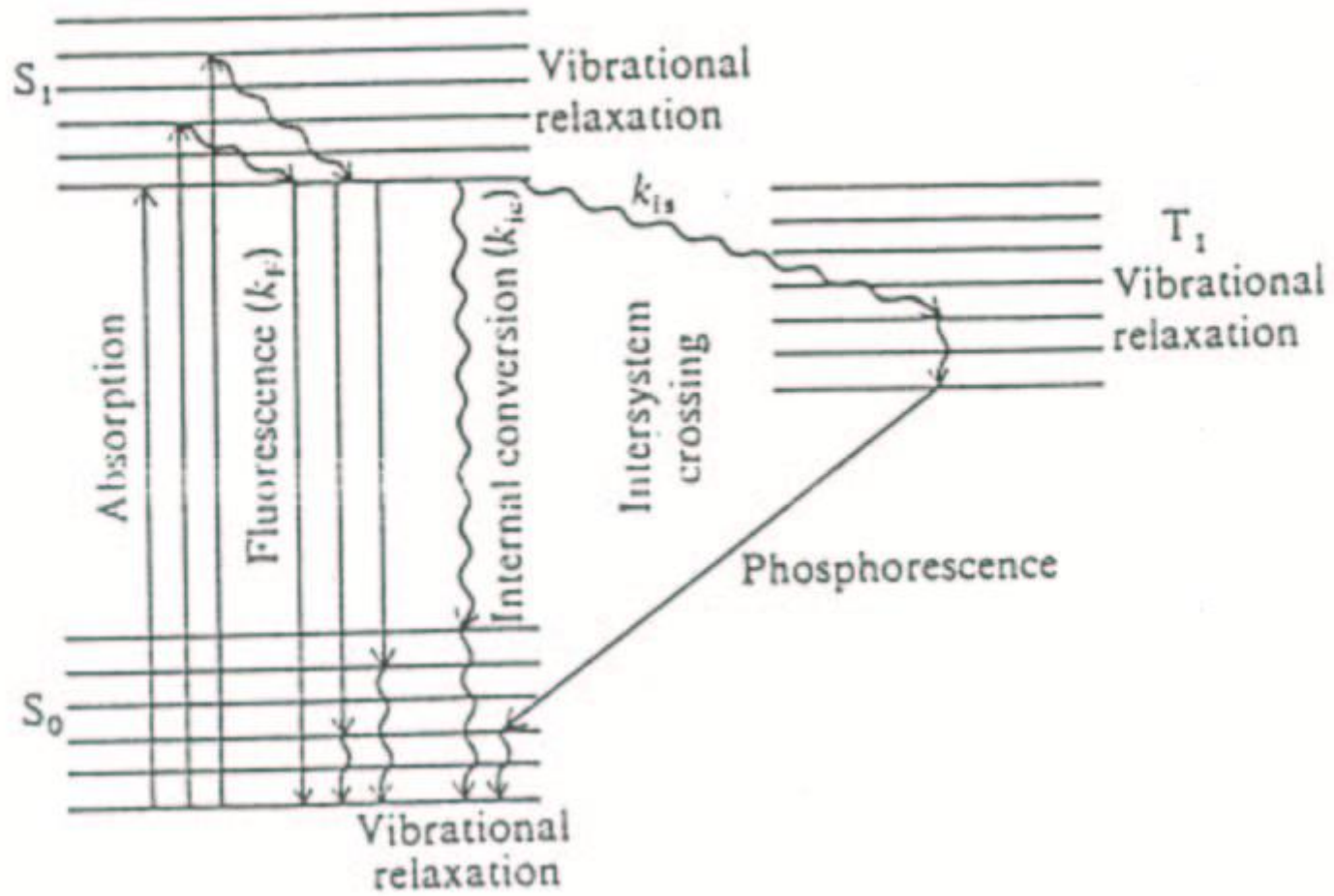
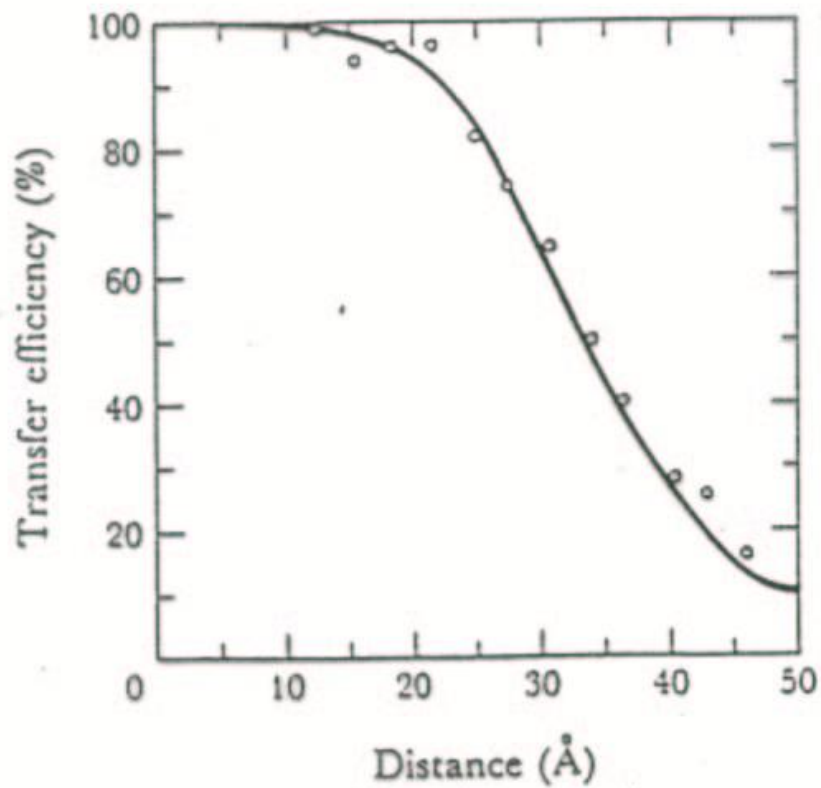
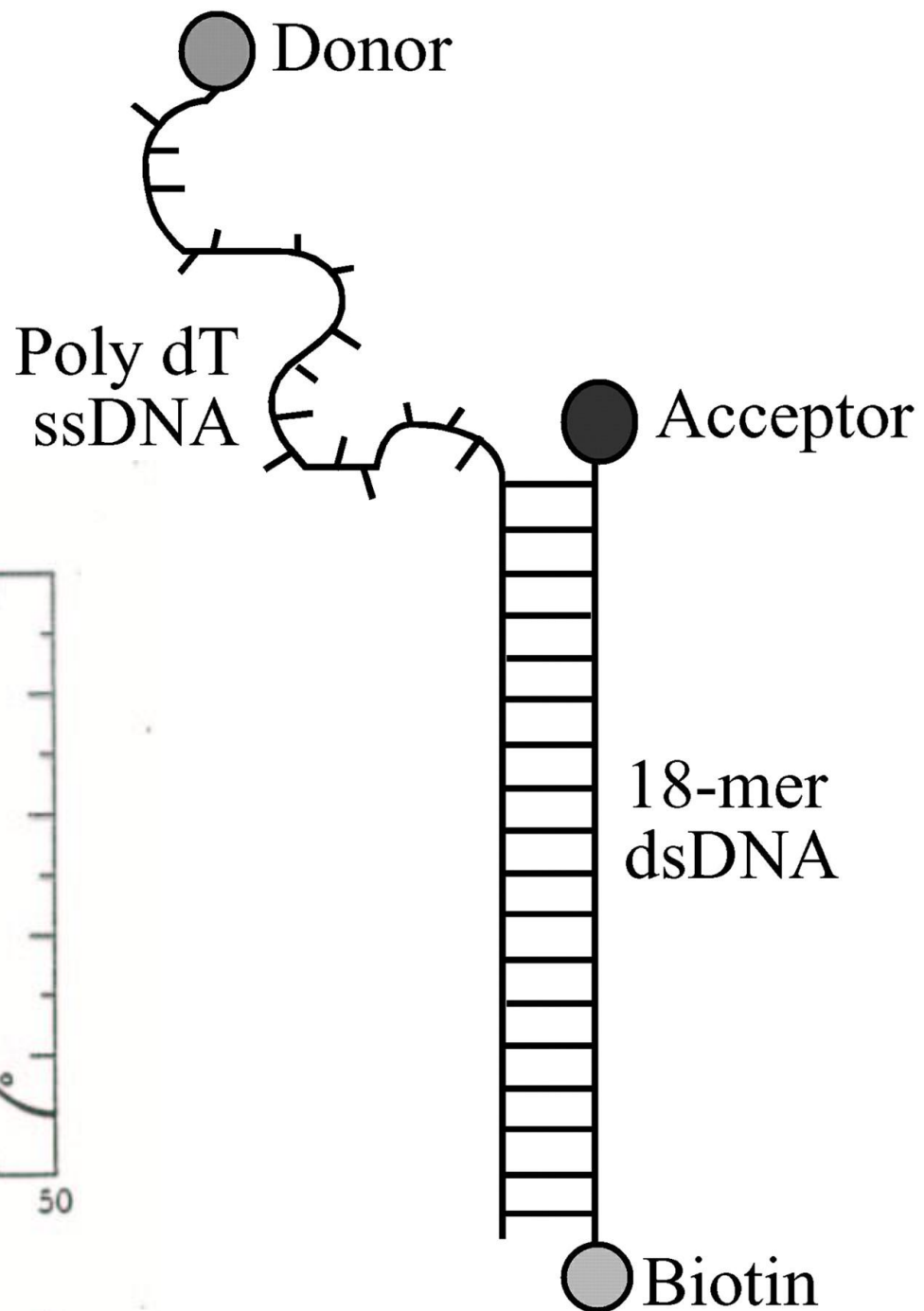


Fluorescence

- Absorption of light occurs within $\sim 10^{-15}$ seconds, leaving a molecule in an excited state
- What happens next?
 - If no photon is re-emitted, the molecule probably loses the energy via a collision with solvent molecules
 - If a photon is emitted then it can be of several types:
 - Scattered at the same frequency/energy
 - Fluorescent at a longer wavelength (takes \sim ns)
 - Phosphorescent – similar to fluorescence but transition is from a triplet state (with electrons parallel $\uparrow\uparrow$; fluorescence is from a singlet state with paired $e^-\uparrow\downarrow$) (takes $>$ msec)
 - Resonant energy transfer (FRET) – donor and acceptor groups have a common vibrational energy level: $A + hf \rightarrow A^*$; $A^* + B \rightleftharpoons A + B^*$; $B^* \rightarrow B + hf$; A & B must lie close to one another – technique can be used as a “yardstick”

Energy Levels





Quantum Yield

- All of these processes compete with one another
- The quantum yield for fluorescence

$$Q_{\text{fluorescence}} = \frac{\# \text{ fluorescent photons}}{\# \text{ absorbed photons}}$$

Each other process has a Q and all must add up to 1:

$$\sum Q_i = 1$$

Two types of factors affecting $Q_{\text{fluorescence}}$:

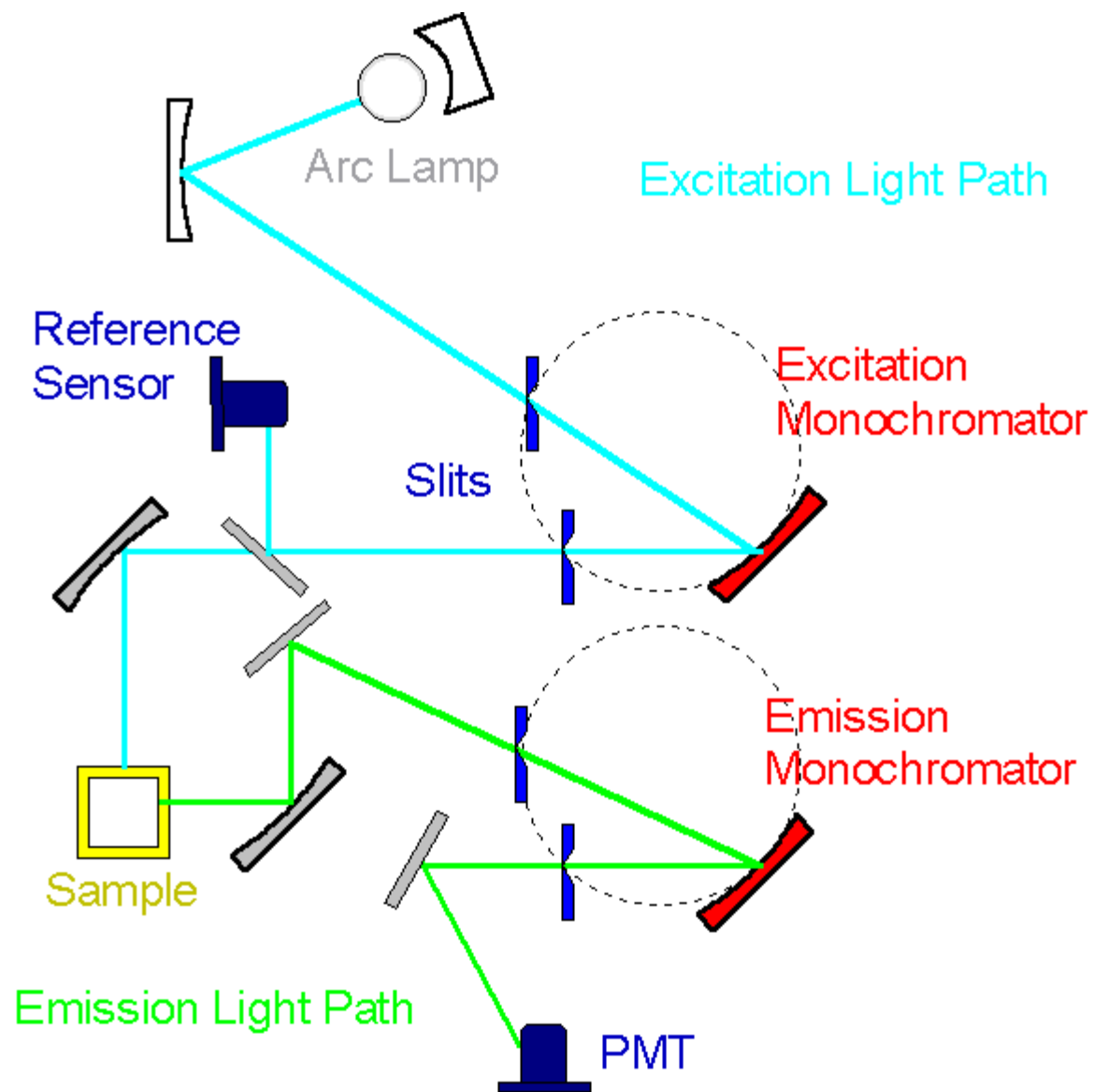
- internal – with more vibrational levels closely spaced (more flexible bonds), fluorescence is more easily quenched, losing energy to heat

best fluors are stiff ring structures: Tryp, Tyr

- environmental factors such as T, pH, neighboring chemical groups, concentration of fluors; generally more interesting

Instrumentation

1. 90° measurement to avoid scattering or direct transmitted beam
2. Very low concentration can be used to keep I_{fluor} linear in concentration
$$I = I_0 Q (1 - e^{-\epsilon c l}) = (\text{for small } c) I_0 Q \epsilon c l = Kc$$
3. Sensitivity is very high since no bkgd signal – no difference measurement (blank) needed as in absorption
4. Measure either I vs λ_{emitted} for a given $\lambda_{\text{inc}} =$ **emission spectrum** OR measure I vs $\lambda_{\text{exciting}}$ at fixed $\lambda_{\text{emitted}} =$ **excitation spectrum**
5. Simple fluorometer uses interference filters for incident & 90° emission – better machines use gratings and scan to get a spectrum



Spectra

Record uncorrected spectra directly –

3 types of corrections needed:

a. Output I_0 of light source varies with λ_{inc}

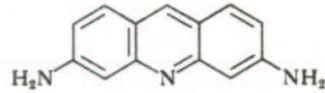
b. Variable losses in monochromators with λ_{inc} or emitted

c. Variable response of PMT with λ_{emitted}

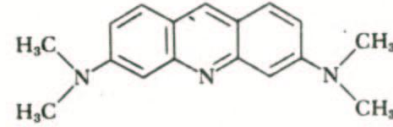
Typically absolute measurements are not done and so no corrections are made – only comparisons

Fluors

- Intrinsic: “chromophore” = e.g. Try, Tyr, Phe – best is Try; I_{fluor} depends strongly on environment
 - Extrinsic: attach fluor to molecule of interest; must:
 - Be tightly bound at unique location
 - Have fluorescence that is sensitive to local environment
 - Not perturb molecules being studied
- Examples: ANS & dansyl chloride fluoresce weakly in water, but strongly in non-polar solvents;
Acridine O used with DNA – green on d-s, red-orange on s-s

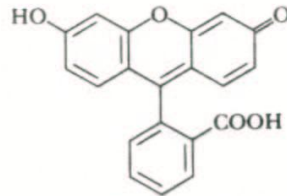


Proflavin

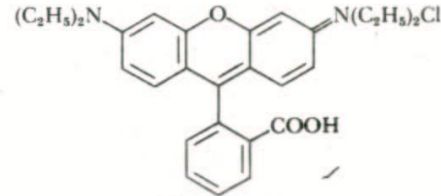


Acridine orange

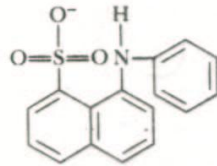
Green on d-s
DNA; red-orange
on s-s DNA



Fluorescein

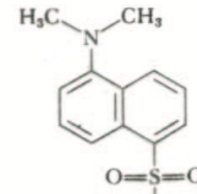


Rhodamine B



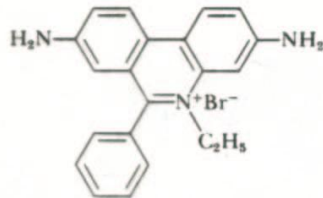
1-Anilino-8-naphthalene
sulfonate (ANS)

Weak in
water; strong
in non-polar
solvents

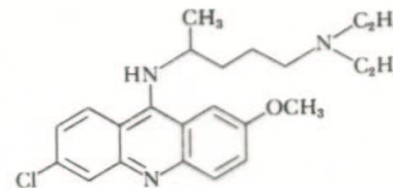


Dansyl chloride

Used with
DNA



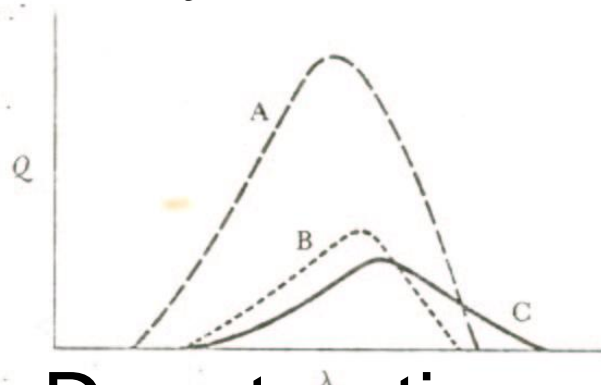
Ethidium bromide



Quinacrine chloride

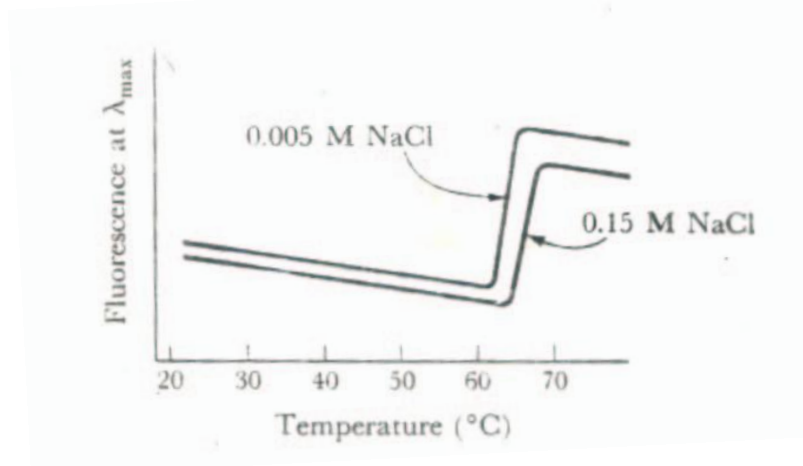
Two Application Examples

1. Detect conformational changes in an enzyme when a co-factor binds



A w/o added co-factor; B with added co-factor; C = free Tryptophan

2. Denaturation of a protein

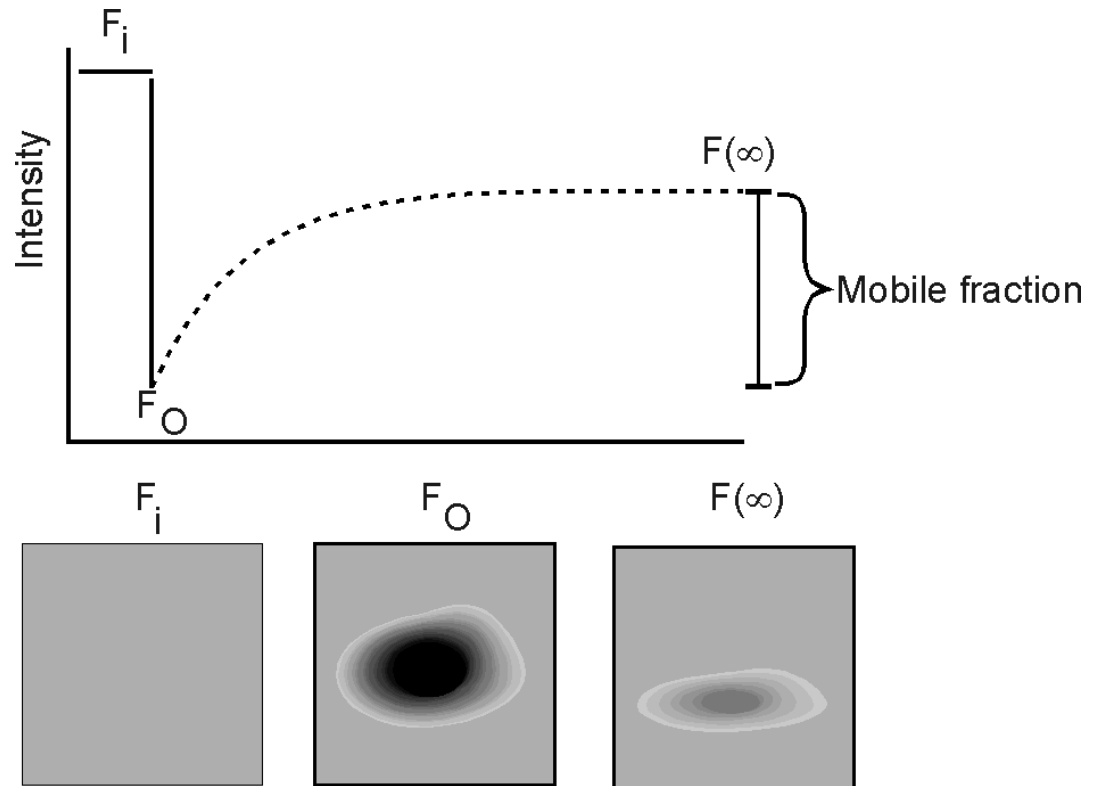


Helix-coil transition of a protein; in 0.15 M NaCl the protein is more stable – higher T needed for transition

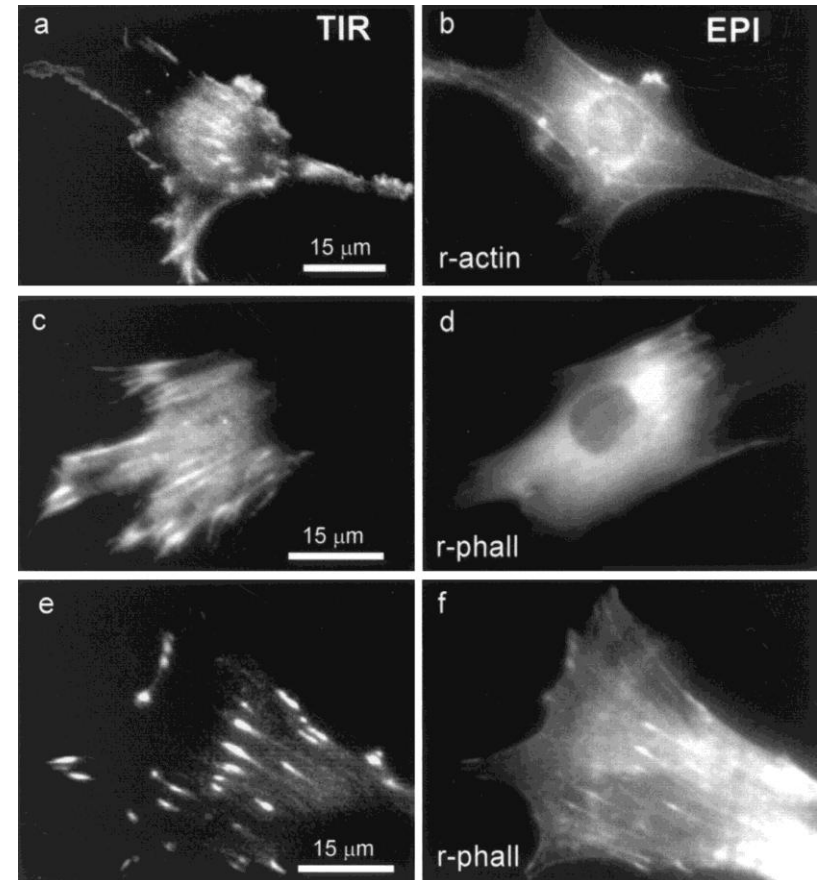
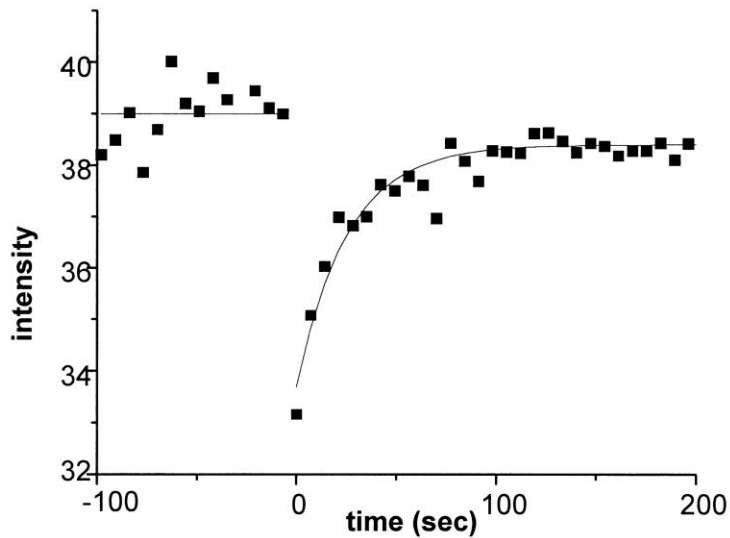
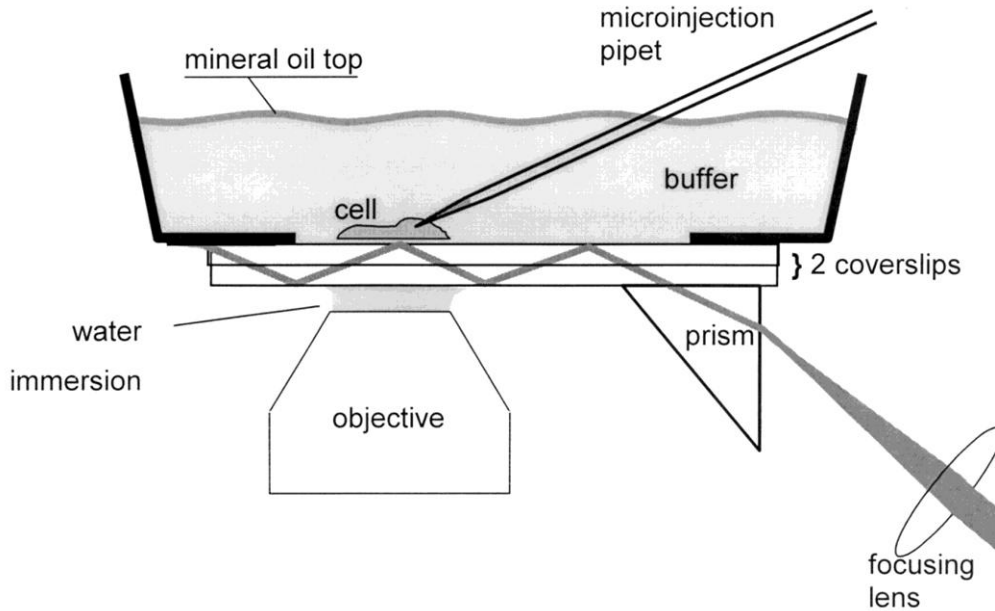
FRAP

- High power bleach pulse
- Low power probe
- Look at 2-D diffusion

$\langle r^2 \rangle = 4Dt \sim$
size² beam
focus



TIR-FRAP



Rhodamine labeled actin/phalloidin